

Synthesis of Edatrexate (2-¹³C-Glutamate)

Joseph I. DeGraw*, William T. Colwell*, and Thomas Jue+

*Bioorganic Chemistry Laboratory
SRI International
Menlo Park, California 94025

+Department of Biological Chemistry
University of California
Davis, California 95616

Summary

The experimental antitumor drug Edatrexate, labeled with 99% ¹³C at the 2-position of the glutamate acid group was required for ¹³C-magnetic resonance spectroscopy studies in biological media. Coupling of 2,4-diamino-4-deoxy-10-ethyl-10-deazapteroic acid with diethyl L-2-¹³C-glutamate as promoted by BOP reagent afforded Edatrexate (2-¹³C-glu) diethyl ester in 60% yield following purification by column chromatography. Saponification by aqueous NaOH in 2-methoxyethanol gave the target molecule in 44% yield or 26% overall.

Key Words: Edatrexate, antitumor, ¹³C-magnetic resonance spectroscopy

Introduction

We have previously reported¹ the synthesis and biological activity of the antifolate drug 10-ethyl-10-deazaaminopterin (Edatrexate, Scheme 1, compound 5). Further clinical studies have demonstrated its efficacy in the treatment of breast and advanced non-small cell lung cancers²⁻⁴. An important property of Edatrexate was its ability to selectively accumulate in tumor cells vs sensitive normal host tissue. Key

repetitive treatment with 2 to 3 equivalents of glutamate ester compared to only one equivalent for the BOP method. The crude reaction product was purified by flash column chromatography to afford the Edatrexate diethyl ester (**4**) in about 60% yield. This material was found to be identical to unlabeled compound **4** prepared in the same manner. The labeled diester **4** was shown to be equivalent to authentic (**4**) by TLC and ¹H-NMR comparisons. Hydrolysis of the labeled diester (**4**) with NaOH in aqueous 2-methoxyethanol afforded Edatrexate (2-¹³C-glu) in 44% yield. The ¹H-NMR spectrum was equivalent to authentic edatrexate.

Experimental

Analtech silica gel GF plates were used for thin layer chromatography. NMR measurements were conducted on a Varian 300XL spectrometer. Preparative chromatography was performed with Baker flash chromatography silica gel.

Diethyl 2-¹³C-L-Glutamate Hydrochloride (**2**)

L-2-¹³C (99%) Glutamic acid (Cambridge isotopes, 301 mg) was added to 2.5 ml of absolute ethanol saturated with dry HCl at 0-5°C. The mixture was stirred at reflux for 3 hours and evaporated *in vacuo*. Another 2 ml of ethanol was added followed by a 20-minute reflux period and evaporation. The residue was treated with 10 ml of ether to give a white gum. The ether was decanted and the gum dissolved in 5 ml of CHCl₃. The solvent was evaporated and the residue treated twice with 10 ml portions of ether. After drying under high vacuum, the semi-solid product weighed 437 mg (89%).

Edatrexate (2-¹³C-Glutamate) Diethyl Ester (**4**)

To a solution of 612 mg (1.8 mmole) of the 2,4-diamino-4-deoxy-10-deazapteroic acid¹ (**3**) in 24 ml of dimethyl formamide (DMF) was added 0.49 ml (3.6 mmole) of triethylamine and 798 mg (1.8 mmole) of benzotriazole oxophosphonate (**BOP**) reagent. The mixture was stirred 1.5 hours at room temperature and 437 mg (1.8 mmole) of diethyl L-2-¹³C-glutamate HCL in 1.5 ml of DMF was added. The mixture

was stirred for 20 hours, the solvent removed *in vacuo*, and the residue thoroughly mixed with 20 ml of water. The aqueous phase was decanted and the thick gum dissolved in 35 ml of CHCl_3 , and washed with 15 ml of water. The CHCl_3 extract was dried over MgSO_4 and evaporated to leave 1.29g of a yellow gum. The material was twice shaken with 20-ml portions of ether followed by decantation and drying *in vacuo* to leave 0.88g. The gum was chromatographed on silica gel with elution by CHCl_3 - MeOH, 95:5, to afford 0.57 g (60%) of a yellow glass following treatment with 15 ml of ether. The thin layer chromatogram showed a single U.V.-absorbing spot at R_f 0.4, equivalent to the unlabeled diester prepared by the same procedure. The unlabeled diester was shown to be equivalent to authentic material previously synthesized by comparison of $^1\text{H-NMR}$ spectra. $^1\text{H-NMR}$ (CDCl_3), δ ppm, 0.78(3H,t, CH_3), 1.25(6H,dt, OCH_2CH_3), 1.72(2H,m, CH_2CH_3), 2.10(2H,m,glu-3- CH_2), 2.45(2H,bs,glu-4- CH_2), 3.10(4H,bs,C-9H,C-10H), 4.18(4H,dq, $\text{OC}_2\text{H}_4\text{CH}_3$), 4.80(1H,m,NH), 6.05(1H,bs, NH_2), 6.75(1H,bs, NH_2), 7.15(2H,d,3',5'-H), 7.72(2H,d,2',6'-H), 8.35(1H,s,C-7H). The 2- and 4- NH_2 typically integrate to 1H in CDCl_3 .

Edatrexate (2- ^{13}C -Glutamic acid) (5)

The ^{13}C -labeled diester (4) above (0.57g) was dissolved in 2 ml of 2-methoxyethanol followed by addition of 4 ml of 1.25 N NaOH. The mixture was stirred at room temperature for 2.75 hours and the resulting yellow solution was adjusted to pH 7-8 with glacial acetic acid. The solvent was removed *in vacuo* and the residue was dissolved in 20 ml of water followed by treatment with acetic acid to precipitate the product at pH 5-6. After one hour the yellow crystalline precipitate was collected, washed with water, and dried to afford 224 mg (44%) of product. Edatrexate $^1\text{H-NMR}$ ($\text{NaOD-D}_2\text{O}$), δ ppm, 0.71(3H,t, CH_3), 1.68(2H,m, CH_2CH_3), 1.95 and 2.08(2H,m,glu-3- CH_2), 2.23(2H,m,glu-4- CH_2), 2.99(2H,m,C-9H), 3.14(1H,m,C-10H), 4.23(1H,q,glu-2- CH), 7.11(2H,d,3',5'-H), 7.60(2H,d,2',6'-H), 8.20(1H,s,C-7H). Except for the signal at 4.23 ppm, corresponding to the C-2 glutamate proton, the ^1H NMR spectra of ^{13}C -C2

glutamate labeled edatrexate are identical to the native molecule. Every peak and multiplet structure match precisely. The ¹H spectra of the ¹³C labeled molecule display a residual, < 0.1 proton intensity, ¹³C-¹H glutamate signal at 4.23 ppm, but symmetrically displaced upfield and downfield by 70 Hz at 4.06 and 4.41 ppm are the ¹³C₂-¹H glutamate signals. Such splitting is consistent with the heteronuclear ¹³C-¹H J coupling. Comparing the integrated area of the ¹³C satellites and the ¹³C₂-¹H glutamate signal reveals that the ¹³C enrichment is >90%.

References

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